

Hanyang University NEMPL

Acidic Cleaning Solutions for Post InGaAs CMP Cleaning

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April 11, 2018





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05 Summary



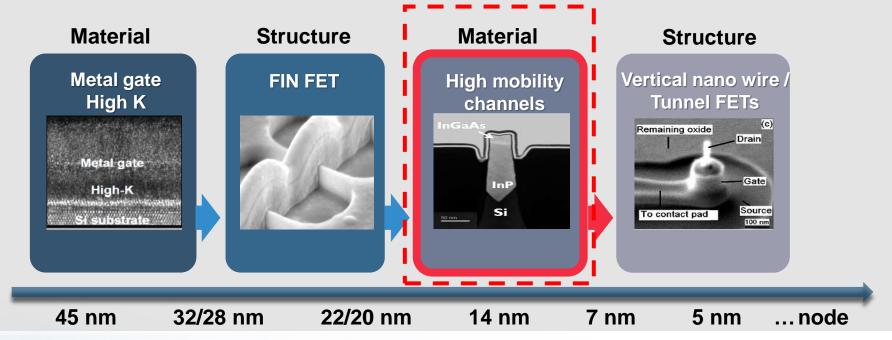
High mobility semiconductor for Logic Device



Technology development trend of semiconductor industry

- ✓ Demand of high speed
- ✓ Decreasing the power consumption
- ✓ Shrinking device size (high density of integration)

Logic Devices Roadmap

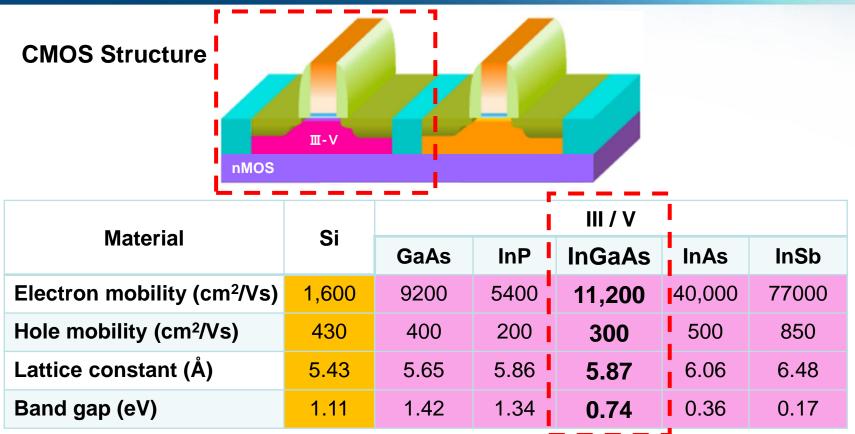


Recently new material is required for high mobility channels

Babu, Suryadevara. Advances in Chemical Mechanical Planarization (CMP). Woodhead Publishing, 2016.

Properties of semiconductor materials



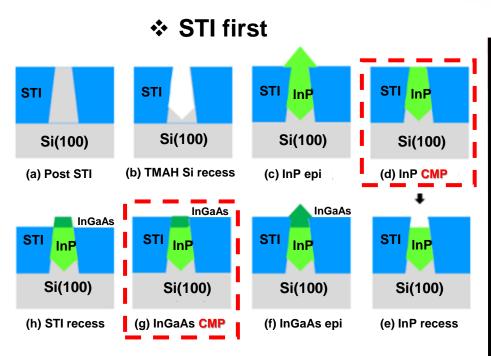


- ➤ Combined elements from group III and V shows higher electron mobility and it is considered as an alternative to Si as a channel material, specifically for nMOS.
- Considering the lattice mismatches and band gap, InGaAs may consider as a promising material for nMOS.

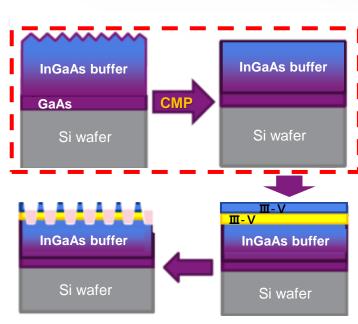
http://www.yokoyama-gnc.jp/english/research/cmos.html

CMP process for III-V Fins





- ✓ ART (Aspect-Ratio Trapping) : Trapped defects in the III-V layer
- CMP is needed to reduce the overburden post epitaxial growth



STI last

- ✓ SRB (Strain Relaxed Buffer):

 Reduced lattice mismatch with Si
- ✓ CMP is needed before epi growth of new layer

In both approaches, InGaAs requires at least 1 CMP step using silica slurry

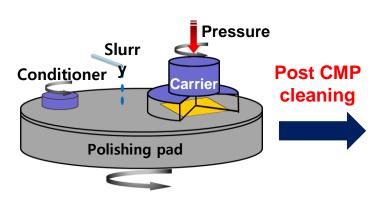
Motivation

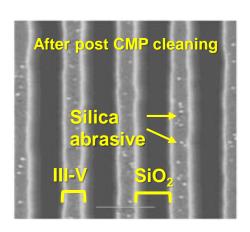


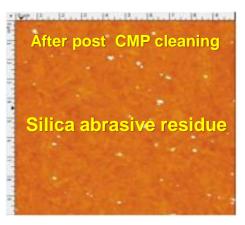
 After CMP process, post InGaAs CMP cleaning process was performed using megasonic, brush and diluted ammonia solutions to remove slurry contaminants.

Issues of post InGaAs CMP cleaning

• The **conventional post InGaAs CMP cleaning process is not effective** for removing nano silica slurry particles from InGaAs surface.







- Development of new cleaning chemical is required for post InGaAs CMP process
 - √ High cleaning efficiency
 - ✓ Minimal material loss with less oxidized surface

Lieve Teugels et al., Semicon Korea 2015, 4-6 February 2015

Experiments for etch rate and oxidation



Materials & Experimental condition

- Substrate: Polished InGaAs (In_{0.53}Ga_{0.47}As; ~75nm thickness)
- Cleaning Chemicals: HCI (0.001 M, 0.005 M, 0.01 M, 0.05 M)

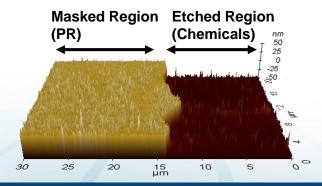
H₂O₂ (0.001 M, 0.005 M, 0.01 M)

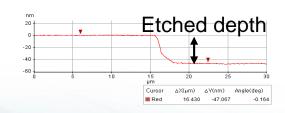
Etch Rate (Material Loss)

- Photoresist (PR) line patterns of 110 μm
 - Selective Etching
- AFM (XE 100, Park System, Korea)
 - Etched step height

Composition Analysis (Oxidation)

- XPS (Sigma Probe, Thermo, UK)
 - Oxide percentage



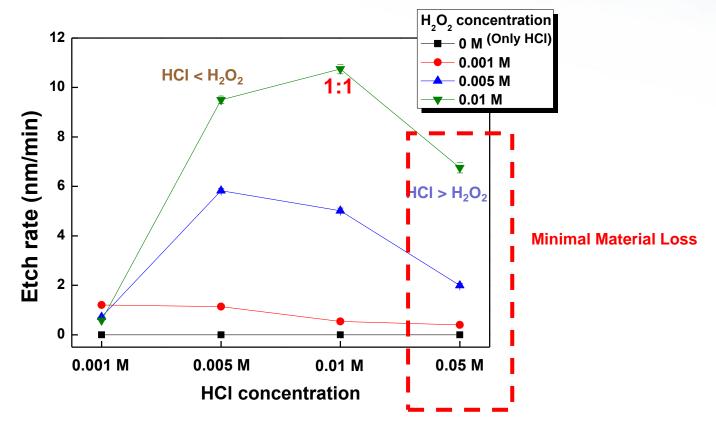


Etch rate
$$(\frac{\mathring{A}}{min}) = \frac{\text{Etched step height}}{\text{Etching time}}$$

Etch rate analysis



Etch rate vs. various concentration of chemical



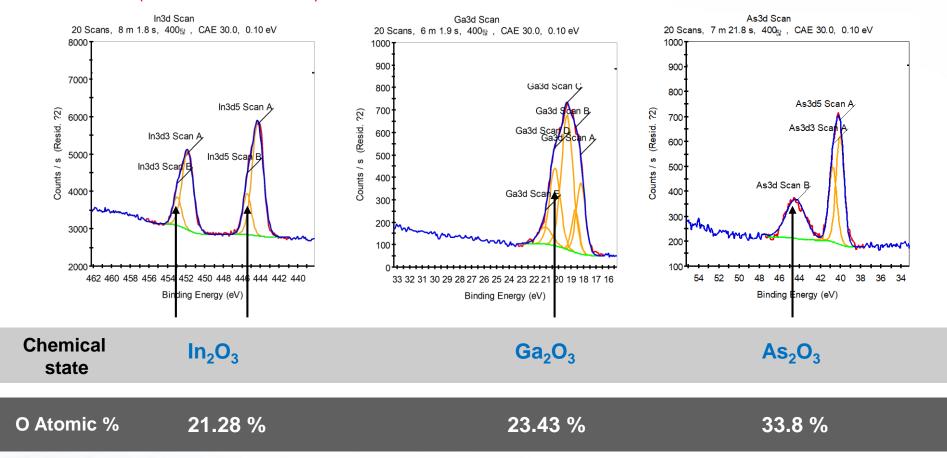
- Only HCl doesn't etch the InGaAs surface without H₂O₂
- \rightarrow Highest etch rate was achieved at HCl: $H_2O_2 = 1:1$
- Lower concentration of HCl than H₂O₂ (ER ↑)
- → Higher concentration of HCl than H₂O₂ (ER ↓)

XPS analysis for oxide content



* XPS analysis was carried out for analyzing the oxide percentage of InGaAs

✓ Ref. (As-received wafer)

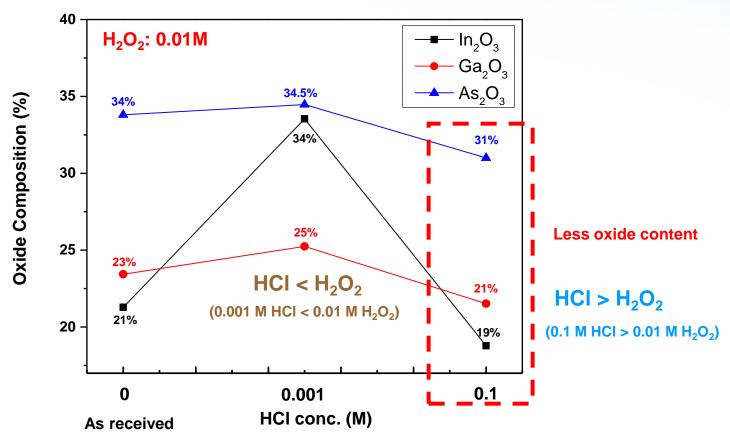


> XPS was used to evaluate the cleaning chemical effect on InGaAs oxidation

XPS analysis for oxide content



Oxide content vs. HCI Conc.



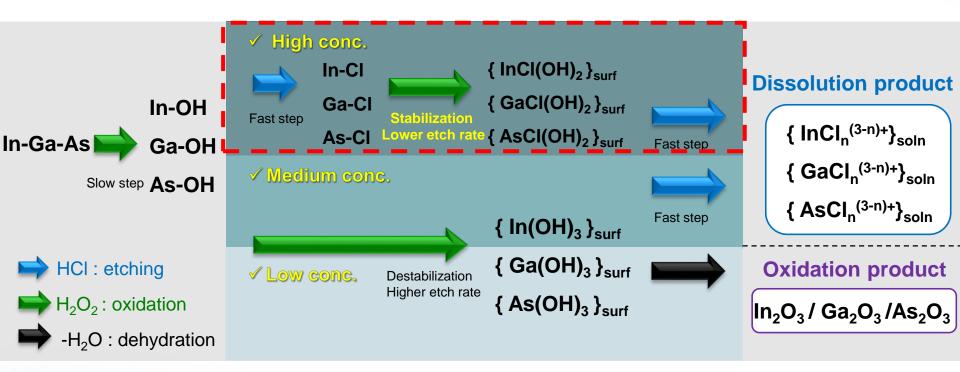
- Lower conc. of HCI increases the oxide composition
- Higher conc. of HCI decreases the oxide composition
- \rightarrow High conc. of HCl than H_2O_2 can minimize the oxide layer formation

Etching mechanism



Etching mechanism according to the HCl concentration

- ✓ High HCl concentration: oxidation rate < dissolution rate (ER ↓) / (Oxidation ↓)
 </p>
- ✓ Low HCl concentration : oxidation rate > dissolution rate (ER ↑) / (Oxidation ↑)



 \triangleright Based on the material loss and surface oxidation results, higher conc. of HCI (0.05 M) and lower conc. of H₂O₂ (0.001M) is considered as a optimized cleaning solution

Experiments for the evaluation of PRE



Materials & Experimental condition

- Substrate : InGaAs (In_{0.47}Ga_{0.53}As)
- Particle: Silica (130 nm, 289 nm)
- Optimized Conc. of cleaning solution: HCI (0.05M) + H₂O₂ (0.001M)
- **Surfactant : ADS** (Ammonium Dodecyl Sulfate) (0 0.5mM)
- Megasonic: batch type, 1MHz, 600W

Particle contamination

- Particle contamination
 - Wet deposition
 - 2 x 10⁻⁵ wt% Silica particle solution

Cleaning process

- Chemical cleaning
 - Only Cleaning Chemical
 - Cleaning Chemical + ADS
- Physical cleaning
 - Cleaning Chemical + MS
 - Cleaning Chemical + ADS + MS

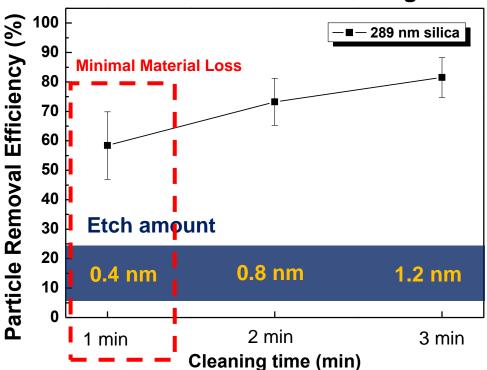
PRE Analysis

- Optical microscope (LV100D, Nikon, Japan)
 - Particle counting (289 nm)
- FE-SEM (MIRA3, TESCAN, Czech)
 - Particle counting (130 nm)

PRE vs. Etch amount



❖ PRE and Etch amount vs. Cleaning Time



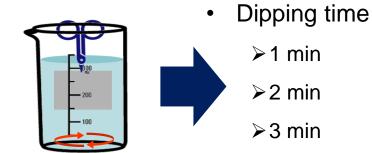
Cleaning condition

Cleaning solution :

 $HCI: 0.05M, H_2O_2: 0.001M$

pH: 1.36

Dipping process



- ➤ Cleaning time

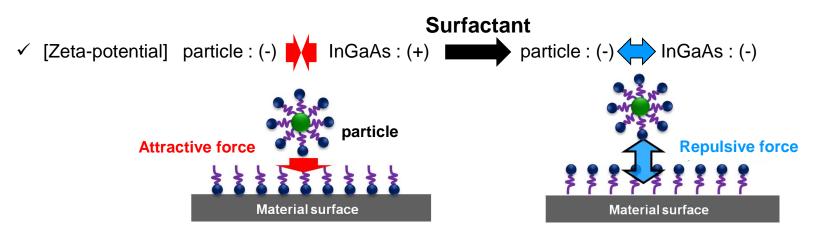
 Etch amount
- > PRE is increased with increase in the etch amount
- Modification of zeta-potential is required to increase the PRE with minimal material loss.
 - ✓ [Zeta-potential] particle : (-) , InGaAs : (+) particle : (-) , InGaAs : (-)

Mechanism of surfactant cleaning



Modification of electrostatic force using surfactant

Introduction of strong electrostatic repulsion between particle and substrate can enhance the PRE during the post CMP cleaning process, which was achieved by using surfactant



Ammonium Dodecyl Sulfate (ADS)

ADS is one of an anionic surfactant widely used for cleaning process

< ADS Properties >

Chemical Formula : C₁₂H₂₉NO₄

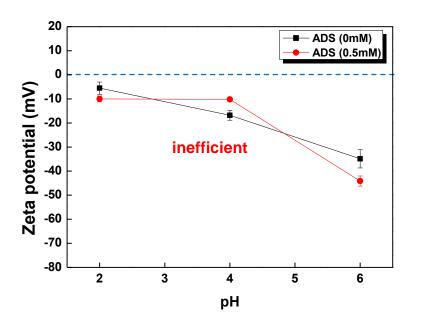
CMC value: 6.6 mM

R. Vos, M. Lux et al., Journal of The Electrochemical Society, 148 (12) G683-G691 (2001)

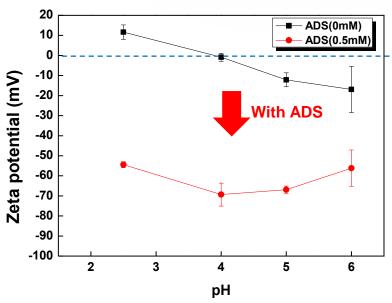
Zeta potential Measurement



Particle zeta-potential (silica)¹



Surface zeta-potential (InGaAs)²



Analysis equipment

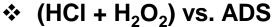
ELS-Z (Electrophoretic Light Scattering Spectrophotometer) Otsuka, Japan
SurPASS Electrokinetic Analyzer Anton Paar, Austria

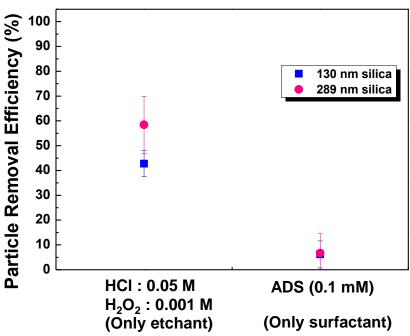
Analysis equipment

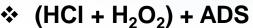
- Zeta potential of InGaAs surface changed significantly after addition of ADS
- > Addition of ADS introduces strong repulsive force between silica particle and InGaAs substrate
 - ✓ Silica particles can be easily removed from the InGaAs surface

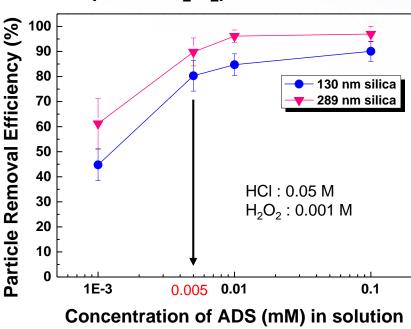
PRE vs. ADS conc.







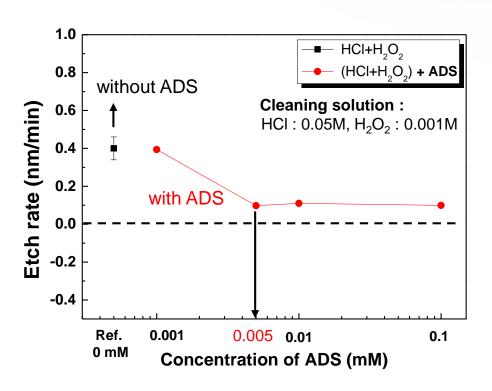




- Using ADS alone is ineffective to remove particles without etchant
 - ✓ Etching process is necessary to remove the particles
- PRE increases dramatically with the addition of ADS
 - ✓ Combined effect of surface etching and electrostatic repulsive force increases the PRE
 - ✓ PRE was increased drastically at the addition of 0.005 mM conc. of ADS

Etch rate vs. ADS conc.



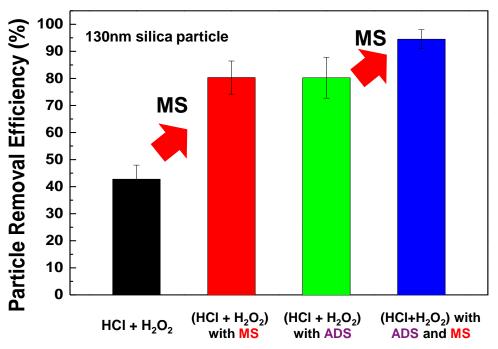


Addition of ADS decreases the etch rate of InGaAs

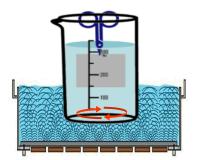
- ✓ Etch rate was decreased drastically at 0.005 mM and above conc. of ADS
- ✓ Introduction of ADS not only increasing the PRE but also decreasing the material loss

PRE vs. Megasonic





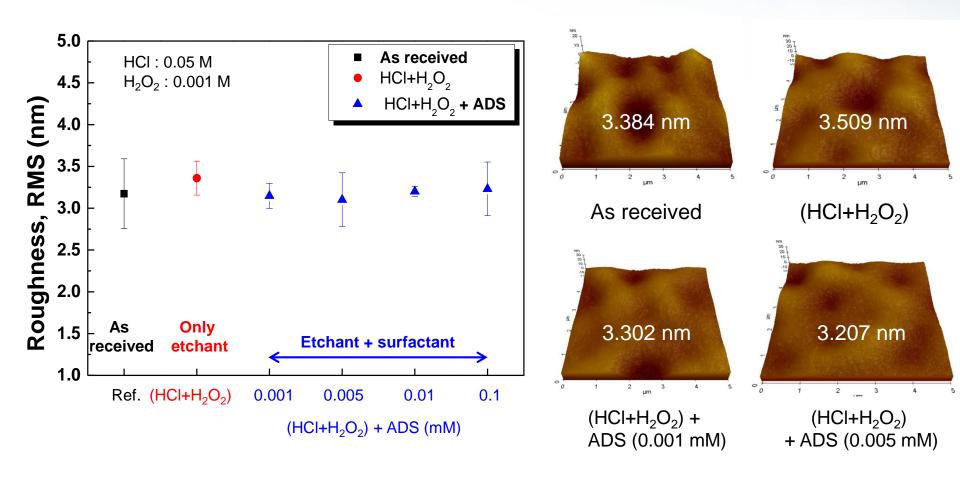
- Cleaning condition
 - ✓ Cleaning solution : HCI : 0.05M, H₂O₂ : 0.001M ADS (0.005 mM)
 - ✓ Megasonic (1 MHz, 600 W)



- Megasonic increases the PRE of various cleaning solutions
- ➤ HCl and H₂O₂ cleaning solution with ADS and MS achieved above 95% PRE
 - ✓ Mechanism of megasonic cleaning : acoustic streaming, bubble collapse
 - ✓ Combined effects of physical force, surface etching and electrostatic repulsive force

Surface roughness





- After cleaning process, there is no significant change in roughness compared to the initial state
 - ✓ Etch amount was negligible to affect the roughness (minimized surface damage)

Summary



- * Acid cleaning solution was used for post InGaAs CMP cleaning process.
- ❖ Higher HCl conc. than H₂O₂ is considered as a optimized cleaning solution based on the results of material loss and surface oxidation.
- **❖ ADS** was introduced for **increasing repulsive force** between silica particle and InGaAs substrate to **enhance the PRE**.
- Introduction of ADS not only increasing the PRE but also decreasing the material loss.
- ❖ There was no noticeable roughness change was observed after cleaning process compared to the initial state.
- **❖** Acid cleaning with surfactant and megasonic achieved above 95% PRE.

